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CHAPTER 7 REVIEW Chemical  
Formulas and Chemical Compounds  
SECTION 1 SHORT ANSWER Answer  
the following questions in the space  
provided. 1. c In a Stock system name  
such as iron(III) sulfate, the Roman  
numeral tells us (a) how many atoms of Fe

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are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

7 Chemical Formulas and Chemical  
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Formulas and Chemical Compounds

MIXED REVIEW SHORT ANSWER

Answer the following questions in the  
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following compounds:  $\text{CuCO}_3$  a.

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copper(II) carbonate  $\text{Na}_2\text{SO}_3$  b. sodium sulfite  $(\text{NH}_4)_3\text{PO}_4$  c. ammonium phosphate  $\text{SnS}_2$  d. tin(IV) sulfide  $\text{HNO}_2$  e. nitrous acid 2.

7 Chemical Formulas and Chemical  
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SECTION 1 SHORT ANSWER Answer

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the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain

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5 The Periodic Law

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