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Chapter 18 - Reaction Rates and Equilibrium

a state of balance in which the rates of the forward and reverse reactions are equal; no net change in the amount of reactants and products occurs in the chemical system (18.2)

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The rate of a chemical reaction is expressed as the amount of reactant changing per unit time. Work Step by Step The rate of a chemical reaction is expressed as the amount of reactant

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- The time needed to complete a chemical reaction can vary tremendously.
- Rates measure changes that occur within intervals of

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time. • In chemistry, rates or speeds are expressed in chemical terms. Atoms, ions, and molecules must collide in order to react. Atoms, ions, and molecules can form a chemical bond when they collide, as long as the particles have enough kinetic energy and have the proper orientation.

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equilibrium. STUDY. PLAY. rate.

describes the change over an interval
of time. collision theory. atoms, ions,
and molecules can react to form
products when they collide, provided
that the particles have enough kinetic

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Fraser. How is the rate of a chemical change expressed? in chemistry, the rate of chemical change or the reaction rate is usually expressed as the amount of reactant changing per unit time.

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